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to be successful. As understood, endowment does not suggest that you have wonderful points.

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Gravity. Created by.
jaysleichter. Terms in
this set (49) Valence
electrons. The
electrons that have the
highest energy in an
atom. They are
involved in chemical
bonding. The number
of valence electrons in
each atom helps
determine the
chemical ...

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Atoms and Bonding
Chapter Test A Multiple
Choice Write the letter
of the correct answer
on the line at the left.

- ___ 1. Which is a property shared by most molecular compounds? a. high boiling point b. high melting point c. low melting point d. nonpolar bonds ___ 2. When an atom loses an electron, it becomes a a. positive ion. b.

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negative ion.

**Atoms and Bonding -
Bridgeway**

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Chapter Test A Atoms
and Bonding Using
Science Skills Use the
portion of the periodic
table below to answer
the following
questions. 26.

Classifying A new
period of very heavy
elements from outer
space has been
discovered by

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scientists. In which group does each element belong? Write the element in the appropriate square of period eight.

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Created by. bonniboo.

Terms in this set (67) a

charged group of

covalently bonded

atoms that tends to act

like a single atom is a.

polyatomic atom. a

charged particle is

referred to as. an ion.

Atoms and Bonding

Chapter 5

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Chapter 5 Atoms and
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Preview Questions 4.

Compounds are formed
by a. combining two or
more different

elements. b.

bombarding atoms
with high-speed

particles. c. combining

two or more different

nuclei. d. dissolving a

solid in a liquid.

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Test. PLAY. Match.
Gravity. Created by.
pieface108. All about
atoms and how they
bond. Terms in this set
(18) . Prentice Hall

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Chapter Test
Physical Science:

Chapter 5 Atoms and
Bonding/ 33 Terms.

Emma_Oster9.

chemical bonding 20
Terms.

Elizabeth_Wood6. Ryan
Sturisky science 4.1

4.2 4.3 17 Terms.

sturiskyryan.

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ionic bond. formed

when one or more
electrons are

transferred from one
atom to another ...

compounds made up of
carbon, hydrogen, and
oxygen atoms that are
the main source of
energy for living things

...

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